Roll No.

(07/22-II)

5190

B. Sc. EXAMINATION

(For Batch 2014 & Onwards)

(Second Semester)

CHEMISTRY

Paper-VI (CH-106)

Organic Chemistry

Time: Three Hours

Maximum Marks: 27

Note: Attempt Five questions in all, selecting two questions from each Section. Q. No. 1 is compulsory.

- 1. (a) Why trans alkenes are more stable than cis alkenes?
 - (b) Discuss relative reactivities of alkenes.
 - (c) Define aromaticity and Huckel's rule. Will cyco-octatetraene show aromatic character?

- (d) Chlorine deactivates the ring and directs the substituents to ortho- and parapositions. Justify.
- (e) Give various classes of dienes along with their structures. Out of these which class is more stable and why?
- (f) Acetylene forms metal acetylides but dimethyl acetylene does not form such derivatives, why?
- (g) Alkyl chloride is more reactive than vinyl chloride. Explain. $7 \times 1 = 7$

Section A

- 2. (a) State and explain Markownikoff's rule. 1
 - (b) Mechanism of dehydrohalogenation of alkyl halides.
 - (c) On reductive analysis, a compounds gave the following:
 - (i) Ethanediol
 - (ii) Propanone
 - (iii) Ethanol.

Write the formula of the compound and give its IUPAC name.

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- 3. (a) Discuss the mechanism of Friedal Craft's acylation of benzene. 2
 - (b) What products would you expect to be formed when the following are subjected to nitration: Nitrobenzene, Phenol, Toluene.
- 4. (a) Give the mechanism of sulphonation of benzene.
 - (b) Give suitable names to the following compounds:

(c) Cyanide group is meta directing, explain.

Section B

5. (a) What is 4 + 2 cycloaddition reaction?

Discuss with example and give its synthetic utility.

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- (b) Explain why electrophilic addition products are temperature dependent in conjugated dienes.
- (c) Give the structure and names of the products expected from the reaction of the following with 1, 3-Butadiene: 1 2 moles of bromine, 1 mole of HCl.
- 6. (a) Write short notes on the following: 3
 - (i) Acidic nature of 1-Alkynes
 - (ii) Nucleophilic addition reactions of alkynes.
 - (b) Complete the following reactions: 2
 - (i) CH = CH + C2H5OH

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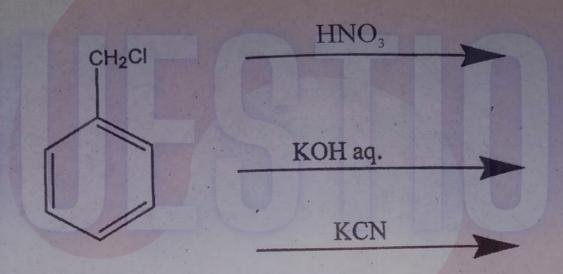
(ii) CH₃ — C = CH

H₂SO₄, HgSO₄

7. (a) Discuss benzyne mechanism for nucleophilic aromatic substitution and give evidence in its favor. 2.5

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(b) Complete the following reactions: 1.5



(c) Vinyl chloride and chlorobenzene show similar reactivity towards SN reaction, justify.