Roll No.

(01/22-II)

5234

B. Sc. EXAMINATION

(Fifth Semester)

CHEMISTRY

CH-303

Organic Chemistry

Time: Three Hours Maximum Marks: 27

Note: Attempt Five questions in all, selecting two questions from each Section. Q. No. 1 is compulsory.

- 1. (a) What is the principle of NMR spectroscopy?
 - (b) What do you understand by peak area?
 - (c) How many kinds of equivalent protons are there in the following:
 - (i) $C_6H_5 CH_3$

- (ii) $CH_3-CH = CH_2$
- (iii) CH₃CH₂CH₂OCH₃
- (iv) CH₃CH₂CHO
- (d) Write a note on shielding and de-shielding of protons.
- (e) How will you distinguish between glucose and fructose?
- (f) What do you understand by the symbols (+), (-), D- and L- used before the names of carbohydrates?
- (g) What are Grignard's reagents? How are they prepared? $7 \times 1 = 7$

Section A

- 2. (a) Distinguish between the following pairs on the basis of NMR data:

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 - (i) CH₃-O-CH₃ and CH₃CH₂OH
 - (ii) CH₃COOC₂H₅ and C₂H₅COOCH₃
 - (iii) CH₃-O- CH₂-CH₃ and CH₃-CH₂-O-CH₂-CH₃.

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- (b) How many signals would be expected in PMR spectrum of each of the following:
 - (i) $C_2H_5COCH_3$
 - (ii) CH₃CHO
 - (iii) CH₃CHBr₂
 - (iv) $C_6H_5 CH_2 CH_3$.
- 3. (a) An organic compound with molecular formula $C_3H_3Cl_5$ gave the following NMR data : 3

 A triplet (δ 4.52, 1H) and A doublet (δ 6.07, 2H)

Assign the structure.

- (b) What is chemical shift? Give the scales of measurement and relation between them.
- 4. (a) Give suitable structures for the compounds with the following structural formula which will give only one PMR signals:
 - (i) C_5H_{12}

- (ii) C₈H₁₈
- (iii) C₂H₆O.
- (b) A compound having molecular formula $C_9H_{11}Br$ gave the following PMR data: Complex multiple of 2H, τ 7.85, Triplet of 2H, τ 7.25, Triplet of 2H, τ 6.62, singlet of 5H, τ 2.78. Assign structure to the compound.

Section B

5. (a) Differentiate between starch and cellulose.

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- (b) What is Osazone? Give mechanism of formation of osazone from fructose. 3
- 6. (a) Why does glucose show that phenomenon of mutarotation? Give its mechanism. 3
 - (b) Give evidence to prove that D(+) Glucose has cyclic structure.

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- 7. (a) What are the products of the reaction of methyl magnesium iodide with each of the following compounds that contain active hydrogen:

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 - (i) Water
 - (ii) Alcohol
 - (iii) Ammonia.
 - (b) What happens when:
 - (i) n-butyl lithium is treated with bromobenzene
 - (ii) phenyl lithium reacts with ethyl bromide?